

f%&t



f%&t

f%&t

CN 108807906 B

f()&t

2020. 12. 15

f%&t 201810602319. X

f) *&t

f%&t 2018. 06. 12

CN 107369825 A, 2017. 11. 21

f*)&t

CN 105990560 A, 2016. 10. 05

CN 108807906 A

CN 107715182 A, 2018. 02. 23

CN 108039500 A, 2018. 05. 15

f()&t 2018. 11. 13

Cheng Lei . Dopamine as the coating

f+&t

agent and carbon precursor for the

453007

fabrication of N-doped carbon coated

46

Fe₃O₄ composites as superior lithium ion

anodes. Nanoscale .2012, 5 1168-1175

f+&t

Dawei Su . Synthesis of tuneable

f+()&t

porous hematites (α-Fe₂O₃) for gas

() 41139

sensing and lithium storage in lithium

ion batteries. Microporous and Mesoporous

f) %&=bh"7 "

Materials .2011, 149 36-45 .

<\$%A (# * (2006. 01)

Peng Sun . Facile synthesis and gas-

<\$%A (# & (2010. 01)

sensing properties of nonodisperse α-

<\$%A (# * & (2006. 01)

Fe₂O₃ discoid crystals. RSC Advances

<\$%A %\$#\$ & (2010. 01)

.2012, 2 9824-9829 .

书1

书3

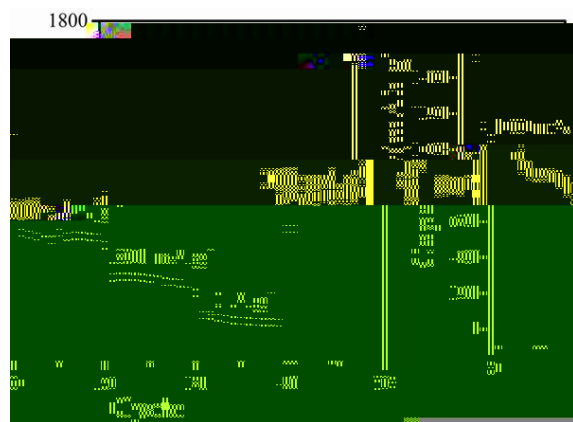
6

f) (&t

f) +&t

Fe₂O₃

Fe₂O₃



Fe₃O₄@N C

CN 108807906 B

1.

1	0.1 g	-Fe ₂ O ₃	30 mL	0.2	0.6 mol L ⁻¹	5
min				160	12 h	
				3	60	2 h
		-Fe ₂ O ₃				
2	0.3 g		-Fe ₂ O ₃	150mL	10 mM Tris	
	10 min	0.15 g		6 h		
	3	60				2 h
	500	600				2 4 h
	Fe ₃ O ₄ @N-C					
2.		1				
		1		0.4 mol L ⁻¹		

[0001]

[0002]

[0003] Fe₃O₄

926 mAh g⁻¹ Fe₃O₄

Fe₃O₄

CN 103227324A
 0.1 A g⁻¹ 20
 454.8 mAh g⁻¹ CN 106450189A
 1C 647 mAh g⁻¹ 100 624
 mAh g⁻¹

[0004]

-Fe₂O₃

-Fe₂O₃

Fe₃O₄@N-C

[0005]

[0006] 1 0.1 g -Fe₂O₃ 30 mL 0.2 0.6 mol L⁻¹
 5 min 160 12 h
 3 60 2 h -

Fe₂O₃

[0007] 2 0.3 g -Fe₂O₃ 150mL 10 mM Tri s

10 min 0.15 g 6 h
 3 60 2 h
 500 600 2 4 h
 $\text{Fe}_3\text{O}_4@N-C$
 [0008] 1 0.4 mol L⁻¹
 [0009] -
 Fe_2O_3 - Fe_2O_3
 $\text{Fe}_3\text{O}_4@N-C$
 [0010] 1 - Fe_2O_3 a b - Fe_2O_3 - Fe_2O_3 -0.2 c
 d - Fe_2O_3 -0.4 e f - Fe_2O_3 -0.6 g h SEM
 [0011] 2 - Fe_2O_3 N₂ - Fe_2O_3 -500 a-c - Fe_2O_3
 - Fe_2O_3 SEM $\text{Fe}_3\text{O}_4@N-C$ d-f Fe_3O_4 -
 0.2@N-C g-i Fe_3O_4 -0.4@N-C j-l) Fe_3O_4 -0.6@N-C m o
 [0012] 3 - Fe_2O_3 -0.4 - Fe_2O_3 -500 $\text{Fe}_3\text{O}_4@N-C$ Fe_3O_4 -0.2@N-C Fe_3O_4 -0.4@N-C Fe_3O_4 -
 0.6@N-C XRD
 [0013] 4 - Fe_2O_3 a - Fe_2O_3 -0.4 b Fe_3O_4 -0.4@N-C c TEM
 Fe_3O_4 -0.4@N-C d Mapping e-h
 [0014] 5 Fe_3O_4 -0.4@N-C 100 mA g⁻¹
 [0015] 6 Fe_3O_4 -0.4@N-C
 [0016]
 [0017] 0.1 g - Fe_2O_3 30 mL 0.2 mol L⁻¹ 0.4 mol
 L⁻¹ 0.6 mol L⁻¹ 5 min 160
 12 h 3
 - Fe_2O_3 - Fe_2O_3 -0.2 - Fe_2O_3 -0.4 - Fe_2O_3 -0.6
 [0018] 0.3g - Fe_2O_3 0.3 g - Fe_2O_3 -0.2 - Fe_2O_3 -0.4 - Fe_2O_3 -0.6
 150mL 10 nM Tri s 10 min 0.15 g
 6 h 3
 - $\text{Fe}_2\text{O}_3@PDA$ - Fe_2O_3 -0.2@PDA - Fe_2O_3 -0.4@PDA - Fe_2O_3 -0.6@PDA

		500	4 h	Fe ₃ O ₄ @N-C	Fe ₃ O ₄ -O. 2@N-C	Fe ₃ O ₄ -O. 4@N-C
Fe ₃ O ₄ -O. 6@N-C		PVDF		7: 2: 1		
[0019]			14 mm		2025	
60 μm						
[0020]				-Fe ₂ O ₃		Fe ₃ O ₄ @N-C
[0021]	1		SEM	1		PDF#
33-0664	-Fe ₂ O ₃	Hemite syn			-Fe ₂ O ₃	
[0022]	2		SEM	2		-Fe ₂ O ₃
	N ₂			2 d-o		
				-Fe ₂ O ₃		2 m
o	-Fe ₂ O ₃					
[0023]	3		XRD	3	-Fe ₂ O ₃	Fe ₃ O ₄ PDF#19-0629
					-Fe ₂ O ₃ -O. 2@PDA	-Fe ₂ O ₃ -O. 4@PDA
					-Fe ₂ O ₃ -O. 6@PDA	
	Fe ₃ O ₄ -O. 2@N-C	Fe ₃ O ₄ -O. 4@N-C	Fe ₃ O ₄ -O. 6@N-C			
[0024]	4		TEM	Mapping	4 a	-
Fe ₂ O ₃						-Fe ₂ O ₃
				4 c		
				Fe ₃ O ₄ -O. 4@N-C	3@N-C	3@N-C

